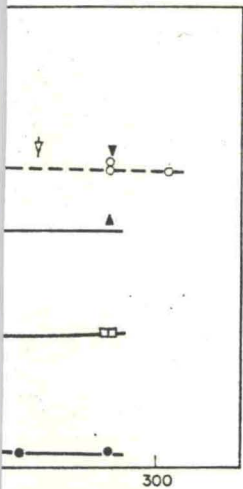


expect a decrease of ρ_{ph} with T much bigger at low temperatures. This is illustrated in the alkali metals.



(3) Resistivity in the alkali metals as a function of temperature (from Phillips, 1965.)

$$(40)$$

and $f(T/\theta)$ is some universal function of temperatures and as T^6/θ^6 at low temperatures (approximately true for several metals, it is true for one metal under independent on pressure) then we can describe its temperature dependence (see Guggenheim, 1962):

$$\frac{\partial \ln \rho_{ph}}{\partial \ln T} = \left(1 + \frac{\partial \ln \rho_{ph}}{\partial \ln T} \right) \quad (41)$$

characterizes the vibrational frequencies of the force constants, i.e., to increase with respect to distance. The effect of pressure is hence to increase its curvature,

In this expression we are treating K and θ and hence their volume derivatives as independent of temperature. Consequently if the electrical resistivity follows a reduced equation of state of the form shown in equation (40), we expect a linear relationship between the logarithmic volume derivative of ρ_{ph} and its logarithmic temperature derivative. This means that where the temperature dependence of ρ_{ph} changes rapidly with temperature the volume dependence will likewise change rapidly.

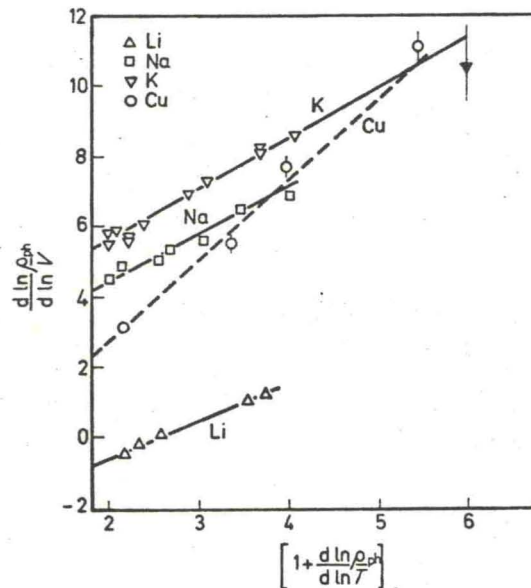


FIG. 21. Relationship between volume coefficient and temperature coefficient of resistivity. (From Dugdale, 1961.)

This relationship has been tested experimentally and the results are shown in Fig. 21 (Dugdale, 1961; Dugdale and Guggenheim, 1962). It is seen that a relationship of this kind does indeed hold. On the other hand we saw that the low-temperature behaviour of the electrical resistivity depended on both the shape of the Fermi surface and on the elastic anisotropy in a way that did not allow them to be separated in any simple fashion. This means that θ in equation (40) does not describe simply the lattice properties of the metal and so the reduced equation of state does not allow the lattice properties to be simply separated out from the electron properties as was originally hoped. The linear relationship in equation (41) is interesting and perhaps